

Structure and High-Pressure Polymorphism of Strontium Metasilicate

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Abstract

Single crystals of α -, δ - and δ' -SrSiO₃ were prepared at pressures of 3.5, 4.0 and 5.5 GPa using KCl as a flux. The atmospheric-pressure phase, α -SrSiO₃, is monoclinic, space group *C*2, with *Z* = 12, *a* = 12.323 (5), *b* = 7.139 (2), *c* = 10.873 (5) Å, β = 111.58 (4)°, and *V* = 889.5 (6) Å³. The two high-pressure phases, δ - and δ' -SrSiO₃, are, respectively, triclinic, space group *P*1̄, with *Z* = 6, *a* = 6.874 (2), *b* = 6.894 (2), *c* = 9.717 (3) Å, α = 85.01 (3), β = 110.57 (3) and γ = 104.01 (2)°, and monoclinic, space group *P*2₁/*c*, with *Z* = 8, *a* = 7.452 (4), *b* = 6.066 (2), *c* = 13.479 (7) Å and β = 117.09 (4)°. Of these phases, the structure of α -SrSiO₃ was determined and refined to *R* = 0.053 for 730 observed reflections with isotropic thermal parameters. The Sr atoms occupy three kinds of sites [Sr(1), Sr(2) and Sr(3)], which are surrounded by eight O atoms with Sr–O distances of 2.39 (1) to 2.83 (1) Å. Three of the distorted SiO₄ tetrahedra with Si–O bond lengths of 1.48 (1) to 1.85 (2) Å form a (Si₃O₉)⁶⁻ ring by sharing O atoms. The structure consists of Sr²⁺ ions and (Si₃O₉)⁶⁻ rings alternately packed along the direction perpendicular to (001).

Introduction

Alkaline-earth metasilicates have been found to show various high-temperature and high-pressure polymorphs (Toropov, Barzakovskii, Lapin & Kurtseva, 1972). At atmospheric pressure, CaSiO₃ crystallizes in α (pseudowollastonite) or β (wollastonite or parawollastonite) forms. The β phase transforms into the α phase above *ca* 1400 K. The structure of β -CaSiO₃ consists of (SiO₃)_∞ chains of SiO₄ tetrahedra (Mamedov & Belov, 1956; Trojer, 1968), while α -CaSiO₃ has been presumed to contain an (Si₃O₉)⁶⁻ ring of three SiO₄ tetrahedra (Jeffery & Heller, 1953). This silicate also forms two high-pressure polymorphs, δ - and ε -CaSiO₃, by treatments at 3–20 GPa and 1170–1770 K (Ringwood & Major, 1967; Liu &

Ringwood, 1975). Trojer (1969) has determined the structure of δ -CaSiO₃, the structural framework of which involves the same (Si₃O₉)⁶⁻ rings as α -CaSiO₃. The ε phase crystallizes in the cubic system (perovskite type) and the structure consists of an (SiO₃)_∞ network of SiO₆ octahedra (Liu & Ringwood, 1975). For SrSiO₃, three polymorphs have been obtained: one atmospheric-pressure phase (α form) and two high-pressure forms (Shimizu, Syono & Akimoto, 1970). It has been confirmed by the X-ray analyses of Hilmer (1962) that α -SrSiO₃ contains (Si₃O₉)⁶⁻ rings in a similar manner as α -CaSiO₃, although the crystals used were not good enough to be able to perform a detailed structure analysis. In this study we have determined the structure of α -SrSiO₃, and characterized its high-pressure polymorphism.

Experimental

Single crystals of SrSiO₃ polymorphs were grown from a mixture of polycrystalline α -SrSiO₃ containing KCl as flux. The SrSiO₃:KCl molar ratio was 10:1, and the pressures applied were 3.5, 4.0 and 5.5 GPa for the α , δ and δ' phases, respectively. For some runs, for samples containing a larger proportion of KCl than the above-mentioned ratio, an unknown phase was formed as a by-product. The high-pressure treatments were carried out with a cubic anvil-type apparatus according to the method previously reported (Shimada, Ogawa, Koizumi, Dacheille & Roy, 1979). The samples were maintained under the appropriate pressures at 1570 or 1670 K for 60 min and then allowed to cool to 1070 K at a rate of 2 K min⁻¹. Transparent single crystals (0.3 mm long) were obtained; the crystal habits were plates for the α and δ' forms, and needles for the δ . Polycrystalline samples of the SrSiO₃ polymorphs were obtained as follows: α -SrSiO₃ was prepared by twice heating a stoichiometric mixture of SrCO₃ and SiO₂ at about 1575 K for 3 h in air, and δ - and δ' -SrSiO₃ were obtained by treatments of α -SrSiO₃ at 4 and 6 GPa at about 1275 K.

The symmetries, space groups and approximate cell parameters of SrSiO₃ polymorphs were determined from oscillation and Weissenberg photographs. From

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Table 1. Additional crystal data for α -SrSiO₃

$M_r = 163.70$
$\lambda = 0.71069 \text{ \AA}$
$\mu(\text{Mo } K\alpha) = 17.83 \text{ mm}^{-1}$
$D_m = 3.64 \text{ Mg m}^{-3}$
$D_x = 3.67$
$F(000) = 912$
Crystal size: $0.20 \times 0.20 \times 0.07 \text{ mm}$

the observed systematic absences (α -SrSiO₃, $h + k = 2n + 1$ for hkl ; δ -SrSiO₃, no conditions; δ' -SrSiO₃, $l = 2n + 1$ for $h0l$ and $k = 2n + 1$ for $0k0$), α - and δ' -SrSiO₃ were found to belong to the monoclinic space groups $C2$, Cm or $C2/m$ and $P2_1/c$, and δ -SrSiO₃ to the triclinic $P1$ or $P\bar{1}$, respectively. Accurate cell parameters were determined by least-squares treatments for the X-ray powder diffraction patterns ($\text{Cu } K\alpha$) of polycrystalline samples calibrated with high-purity Si as an internal standard. The crystal data of α -SrSiO₃ are shown in the *Abstract* and in Table 1. The cell parameters which have been previously reported by Hilmer (1962) (a monoclinic system with $\gamma = 120^\circ$) were not consistent with our result. The intensity data of α -SrSiO₃ were collected on a Rigaku automated four-circle diffractometer with Mo $K\alpha$ radiation monochromated by a graphite plate. The ω - 2θ scan method was employed with a scanning rate of 4° min^{-1} . All possible hkl and $\bar{h}kl$ reflections were collected up to $2\theta = 60^\circ$. Three standard reflections were monitored every 100 reflections and no apparent decay in intensity was detected. The 730 unique reflections with $|F_o| > 3\sigma(|F_o|)$ were considered as observed. The usual Lorentz and polarization corrections were applied, but the absorption and anomalous-dispersion effects for heavy atoms were not considered.

Structure determination

The structure of α -SrSiO₃ was solved by the direct method (*MULTAN* 78; Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978) and refined by the block-diagonal least-squares method (*HBL5-V*; Ashida, 1979), the function minimized being $\sum w(|F_o| - |F_c|)^2$. The space group $C2$ gave a satisfactory result: Sr(1), Sr(2) and Sr(3) atoms occupy the general positions [site 4(c)]. The coordinates of the remaining atoms (Si and O) were determined from the difference Fourier maps. Among them, the coordinates of Si(1), Si(3), O(1) and O(6) could be located at the special sites 2(a) and 2(b) and those of the other atoms at general positions. Isotropic refinements for all atomic parameters gave $R = 0.053$ and $R_w = 0.082$ for 730 observed reflections, where $R = \sum ||F_o| - |F_c|| / \sum |F_o|$

Table 2. Final positional and thermal parameters for α -SrSiO₃

Standard deviations are given in parentheses.				
	x	y	z	$B (\text{\AA}^2)$
Sr(1)	0.0873 (11)	0	0.2494 (9)	0.34 (2)
Sr(2)	0.2455 (16)	0.4780 (19)	0.2458 (18)	0.33 (2)
Sr(3)	0.4128 (11)	-0.0207 (14)	0.2516 (9)	0.42 (2)
Si(1)	0	0.417 (6)	0	0.34 (7)
Si(2)	0.116 (4)	0.790 (5)	-0.001 (4)	0.45 (5)
Si(3)	0	1.075 (7)	$\frac{1}{2}$	0.31 (7)
Si(4)	0.130 (4)	0.702 (5)	0.498 (3)	0.39 (5)
O(1)	0	0.874 (16)	0	0.28 (18)
O(2)	0.042 (12)	0.323 (13)	0.139 (10)	0.74 (16)
O(3)	0.107 (11)	0.532 (14)	-0.009 (9)	0.73 (14)
O(4)	0.130 (11)	0.842 (12)	-0.142 (11)	0.52 (14)
O(5)	0.223 (12)	0.835 (14)	0.139 (11)	0.90 (17)
O(6)	0	0.605 (16)	$\frac{1}{2}$	0.34 (19)
O(7)	0.049 (9)	1.185 (13)	0.635 (9)	0.69 (15)
O(8)	0.110 (11)	0.911 (12)	0.511 (9)	0.41 (14)
O(9)	0.119 (12)	0.659 (14)	0.350 (11)	1.12 (16)
O(10)	0.213 (9)	0.643 (12)	0.630 (9)	0.40 (13)

and $R_w = [\sum w(|F_o| - |F_c|)^2 / \sum w(F_o)^2]^{1/2}$. * The atomic scattering factors for Sr, Si and O were those in *International Tables for X-ray Crystallography* (1974). The weighting scheme $w = (F_m/F_o)^2$ for $F_o > F_m$ ($=30.0$) and $w = 1.0$ for $F_o \leq F_m$ ($=30.0$) was employed. The final positional and thermal parameters are listed in Table 2.

Results and discussion

The interatomic distances and angles of α -SrSiO₃ are summarized in Table 3. Projections of the α -SrSiO₃

* A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 36368 (3 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

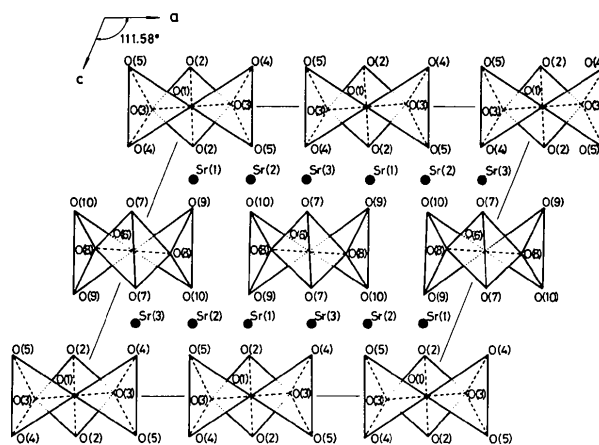


Fig. 1. A projection of the α -SrSiO₃ structure viewed along the b axis.

Table 3. Interatomic distances (Å) and angles (°) in α -SrSiO₃

Standard deviations are given in parentheses.

(a) The Si₃O₉ rings

Si(1) tetrahedron

Si(1)—O(2)	1.56 (1) (×2)	O(2)—Si(1)—O(2')	129.0 (8)
—O(3)	1.59 (2) (×2)	O(2)—Si(1)—O(3)	107.2 (8) (×2)
Average	1.58	O(2)—Si(1)—O(3')	98.6 (8) (×2)
		O(3)—Si(1)—O(3')	117.7 (8)
		Average	109.7

Si(2) tetrahedron

Si(2)—O(1)	1.55 (2)	O(1)—Si(2)—O(3)	110.2 (8)
—O(3)	1.85 (2)	O(1)—Si(2)—O(4)	109.3 (8)
—O(4)	1.65 (1)	O(1)—Si(2)—O(5)	110.2 (8)
—O(5)	1.64 (2)	O(3)—Si(2)—O(4)	101.9 (7)
Average	1.67	O(3)—Si(2)—O(5)	104.7 (7)
		O(4)—Si(2)—O(5)	119.8 (7)
		Average	109.4

Si(3) tetrahedron

Si(3)—O(7)	1.58 (1) (×2)	O(7)—Si(3)—O(7')	120.2 (8)
—O(8)	1.76 (1) (×2)	O(7)—Si(3)—O(8)	104.3 (7) (×2)
Average	1.67	O(7)—Si(3)—O(8')	114.6 (7) (×2)
		O(8)—Si(3)—O(8')	96.7 (6)
		Average	109.1

Si(4) tetrahedron

Si(4)—O(6)	1.75 (2)	O(6)—Si(4)—O(8)	101.9 (7)
—O(8)	1.53 (1)	O(6)—Si(4)—O(9)	101.4 (8)
—O(9)	1.60 (2)	O(6)—Si(4)—O(10)	100.6 (8)
—O(10)	1.48 (1)	O(8)—Si(4)—O(9)	109.0 (7)
Average	1.59	O(8)—Si(4)—O(10)	105.4 (7)
		O(9)—Si(4)—O(10)	133.9 (8)
		Average	108.7

(b) Sr—O distances

Sr(1) dodecahedron

Sr(1)—O(1)	2.68 (2)	Sr(1)—O(7')	2.78 (1)
—O(2)	2.57 (1)	—O(8)	2.83 (1)
—O(4')	2.74 (1)	—O(9)	2.64 (1)
—O(5')	2.67 (1)	—O(10 ⁱⁱⁱ)	2.54 (1)
		Average	2.68

Sr(2) dodecahedron

Sr(2)—O(2)	2.59 (1)	Sr(2)—O(7')	2.80 (1)
—O(3)	2.69 (1)	—O(8 ⁱⁱⁱ)	2.63 (1)
—O(4 ⁱⁱⁱ)	2.42 (1)	—O(9)	2.59 (1)
—O(5)	2.77 (1)	—O(19 ⁱⁱⁱ)	2.70 (1)
		Average	2.65

Sr(3) dodecahedron

Sr(3)—O(2 ⁱⁱ)	2.59 (1)	Sr(3)—O(6 ⁱⁱ)	2.67 (2)
—O(3 ⁱⁱⁱ)	2.59 (1)	—O(7 ⁱⁱⁱ)	2.39 (1)
—O(4')	2.82 (1)	—O(9 ⁱⁱ)	2.69 (1)
—O(5)	2.43 (1)	—O(10 ⁱⁱⁱ)	2.63 (1)
		Average	2.60

Symmetry code: O(*n*ⁱ) (\bar{x}, y, \bar{z}); O(*n*ⁱⁱ) ($\frac{1}{2} + x, \frac{1}{2} + y, z$); O(*n*ⁱⁱⁱ) ($\frac{1}{2} - x, \frac{1}{2} + y, \bar{z}$).

structure viewed along the *b* and *c* axes are shown in Figs. 1 and 2. The structure consists of Sr atoms and (Si₃O₉)⁶⁻ rings, which are alternately pseudo-hexagonally packed along the direction perpendicular to

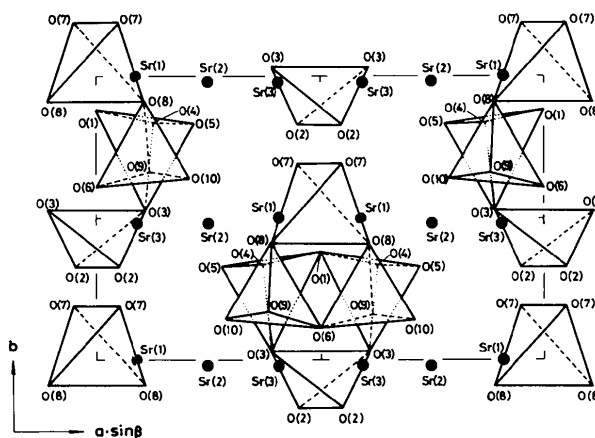


Fig. 2. A projection of the α -SrSiO₃ structure viewed along the *c* axis.

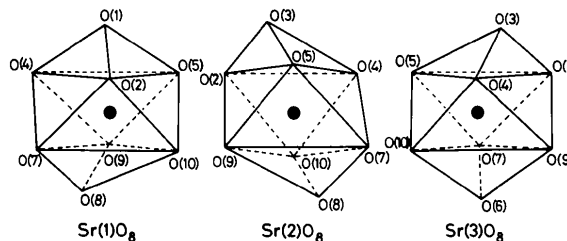


Fig. 3. Schematic illustrations of the SrO₈ polyhedra in α -SrSiO₃.

(001). Because of this arrangement, the symmetry of α -SrSiO₃ has been reported to be pseudo-hexagonal. There are two kinds of (Si₃O₉)⁶⁻ rings, one type comprising one Si(1)O₄ and two Si(2)O₄ tetrahedra and the other one Si(3)O₄ and two Si(4)O₄ tetrahedra with shared O atoms. The atoms Si(1), Si(3), O(1) and O(6) are located on twofold axes parallel to the *b* axis, and the centers of gravity for those (Si₃O₉)⁶⁻ rings also lie on these axes. Consequently, the (Si₃O₉)⁶⁻ rings should show twofold rotation symmetry. The Si—O bond lengths of the SiO₄ tetrahedra vary from 1.48 (1) to 1.85 (2) Å and the O—Si—O angles [96.7 (6)–133.9 (8)°] deviate considerably from the value for a regular tetrahedron (109.47°).

Atoms of Sr occupy the sites Sr(1), Sr(2) and Sr(3). Each Sr is surrounded by eight O atoms, with Sr—O distances from 2.39 (1) to 2.83 (1) Å, to form an SrO₈ dodecahedron (Fig. 3). The polyhedra Sr(1)O₈, Sr(2)O₈ and Sr(3)O₈ are alternately connected with one another to form layers of SrO₈ on planes (004) and (00 $\bar{4}$). The crystal lattice is a three-dimensional arrangement of these layers linked together with the (Si₃O₉)⁶⁻ rings by the O atoms.

Lattice parameters for the polymorphs of CaSiO₃ and SrSiO₃ are summarized in Table 4. Both these silicates have two high-pressure phases in addition to the atmospheric-pressure phase (α form). Among them,

Table 4. *High-pressure polymorphs for CaSiO₃ and SrSiO₃*

The values in parentheses represent the calculated densities (Mg m⁻³).

α Form	δ Form	δ' Form	ϵ Form
CaSiO₃			
$a = 6.90 \text{ \AA}^{(a)}$	$a = 6.695 (5) \text{ \AA}^{(b)}$		$a = 3.485 (8) \text{ \AA}^{(c)}$
$b = 11.78$	$b = 9.257 (7)$		(4.56)
$c = 19.65$	$c = 6.666 (6)$		
$\beta = 90.8^\circ$	$\alpha = 86.63 (5)^\circ$		
(2.90)	$\beta = 76.13 (5)$		
	$\gamma = 70.38 (5)$		
	(3.06)		
SrSiO₃^(d)			
$a = 12.323 (5) \text{ \AA}$	$a = 6.874 (2) \text{ \AA}$	$a = 7.452 (4) \text{ \AA}$	
$b = 7.139 (2)$	$b = 6.894 (2)$	$b = 6.066 (2)$	
$c = 10.873 (5)$	$c = 9.717 (3)$	$c = 13.479 (7)$	
$\beta = 111.58 (4)^\circ$	$\alpha = 85.01 (3)^\circ$	$\beta = 117.09 (4)^\circ$	
(3.67)	$\beta = 110.57 (3)$	(4.01)	
	$\gamma = 104.01 (2)$		
	(3.90)		

References: (a) Jeffery & Heller (1953). (b) Trojer (1969). (c) Liu & Ringwood (1975). (d) This work.

δ -SrSiO₃ is isostructural with the CaSiO₃ analog, where the a , b and c axes of δ -CaSiO₃ correspond to the $-a$, c and b axes of δ -SrSiO₃. On the basis of the structural analysis for δ -CaSiO₃ (Trojer, 1969), the space group of δ -SrSiO₃ can be assigned as $P\bar{1}$ and the structure consists of (Si₃O₉)⁶⁻ rings. Each Sr atom occupies the eight- or six-coordinated site. The crystal lattice of α -CaSiO₃ is also presumed to have a similar arrangement of atoms to that of α -SrSiO₃, but δ' is the phase which has not been observed for CaSiO₃.

The structural framework of ϵ -CaSiO₃ (perovskite type) is a three-dimensional (SiO₃)_∞ network of SiO₆ octahedra. The density of CaSiO₃ considerably increases on changing from SiO₄ to SiO₆ units, *i.e.* on transforming into the ϵ form. From Table 4, the D_x value for SrSiO₃ can be seen not to change appreciably with the phase transformation. This suggests that the structure of δ' -SrSiO₃ consists of anions of SiO₄ tetrahedra in a similar manner to the α and δ forms.

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